

Synthetic Methods

Enantioselective Iridium-Catalyzed Vinylogous Reformatsky-Aldol Reaction from the Alcohol Oxidation Level: Linear Regioselectivity by Way of Carbon-Bound Enolates**

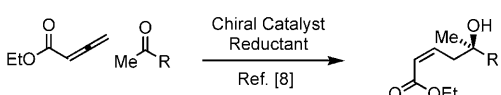
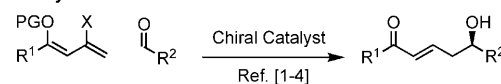
Abbas Hassan, Jason R. Zbieg, and Michael J. Krische*

Vinylogues^[1] of the aldol reaction represent an important class of carbonyl addition processes.^[2] The vast majority of enantioselective vinylogous aldol additions employ dienolates and dienol ethers derived from β -ketoesters^[3] in combination with chiral Lewis acids,^[3a-k] chiral Lewis bases,^[3k-m] or chiral hydrogen-bond donors.^[3n] Dienolates and dienol ethers derived from simple α,β -unsaturated carbonyl compounds^[3k-m,4] and 2-siloxy furans^[5] also participate in enantioselective vinylogous aldol additions. While excellent regio- and enantioselectivities have been obtained in certain cases, the formation and tractability of the requisite dienol ethers pose a barrier to their use. Additionally, chiral Lewis acid catalyzed reactions of silyl dienol ethers often suffer from competing racemic silyl cation catalyzed background reactions, thus requiring slow addition of the dienol ether, cryogenic conditions, and high catalyst loadings. Vinylogous direct aldol additions of unmodified unsaturated carbonyl compounds potentially address these limitations,^[6] however, to date, such transformations are restricted to the use of 2-(5*H*)-furanones or nitriles as aldol donors.^[6,7]

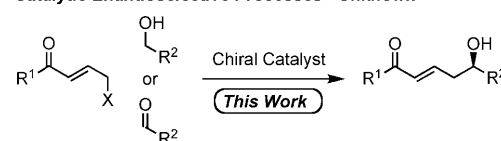
In a significant departure from prior art, Kanai, Shibasaki, and co-workers devised a reductive vinylogous aldol reaction of allenic esters mediated by pinacolborane.^[8] A related reductive process, the vinylogous Reformatsky reaction, could potentially deliver identical products, however, enantioselective variants are unknown.^[9] Herein, under the conditions of C–C bond-forming transfer hydrogenation,^[10,11] we report the first examples of enantioselective vinylogous Reformatsky-type additions, and establish catalytic conditions wherein asymmetric carbonyl addition occurs with equal facility from the alcohol or aldehyde oxidation level (Scheme 1).

In recent work, our research group has found that chiral *ortho*-cyclometalated iridium C,O-benzoates catalyze carbon-

Catalytic Enantioselective Processes - Known

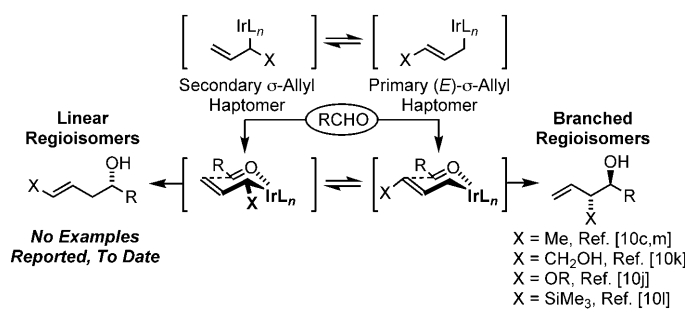


Catalytic Enantioselective Processes - Unknown



Scheme 1. Catalytic enantioselective vinylogous aldol reactions. PG = protecting group.

yl allylation,^[10a,b,e-h] crotylation,^[10c,f] *tert*-prenylation,^[10d,f] and a host of related carbonyl allylation processes. For such “C–C bond-forming transfer hydrogenations”^[10,11] secondary alcohol dehydrogenation triggers reductive generation of allyliridium nucleophiles, which, in the presence of exogenous aldehyde, deliver products of carbonyl allylation. Of greater significance, primary alcohol dehydrogenation can simultaneously generate aldehyde–allyliridium pairs, thus enabling carbonyl addition from the alcohol oxidation level. For all reactions that occur by way of substituted π -allyl iridium intermediates, complete branched regioselectivities are accompanied by good to complete levels of *anti*-diastereoselectivity, thus suggesting carbonyl addition occurs with allylic inversion from the primary (*E*)- σ -allyl haptomer. To date, the formation of linear carbonyl addition products from substituted allyliridium nucleophiles has proven elusive (Scheme 2).

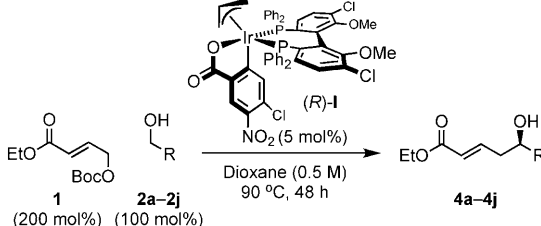


Scheme 2. Carbonyl addition by way of primary σ -allyl haptomers typically generates branched products as *anti*-diastereomers.

[*] A. Hassan, J. R. Zbieg, Prof. M. J. Krische
University of Texas at Austin
Department of Chemistry and Biochemistry
1 University Station—A5300, Austin, TX 78712-1167 (USA)
Fax: (+1) 512-471-8696
E-mail: mkrische@mail.utexas.edu

[**] Acknowledgement is made to the Robert A. Welch Foundation (F-0038), the NIH-NIGMS (RO1-GM069445), the University of Texas at Austin, and the Center for Green Chemistry and Catalysis for generous financial support. The Higher Education Commission of Pakistan is acknowledged for graduate student fellowship support (A.H.).

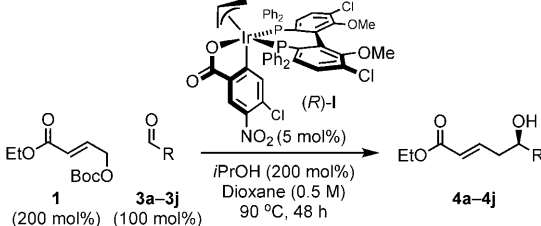
Supporting information for this article is available on the WWW under <http://dx.doi.org/10.1002/ange.201100646>.

Table 1: Enantioselective vinylogous Reformatsky reaction from the alcohol oxidation level.^[a]


Entry	Product	Yield [%]	ee [%]	γ/α
1		87	88	1:2
2		74	94	5:1
3		77	95	5:1
4		59	96	$\geq 20:1$
5		88	96	$\geq 20:1$
6		75	99	$\geq 20:1$
7		70	97	$\geq 20:1$
8		61	90	$\geq 20:1$ ^[b]
9		80	86	$\geq 20:1$
10		80	95	5:1

[a] Yields are of isolated product. Enantiomeric excess was determined by HPLC on a chiral stationary phase. See the Supporting Information for further details. [b] 24 h. Boc = *tert*-butoxycarbonyl, PMB = *para*-methoxybenzyl, Ts = 4-toluenesulfonyl.

It was reasoned that linear carbonyl addition products might be avoided upon stabilization of the secondary σ -allyl haptomer through introduction of a carbonyl moiety, in which case the secondary σ -allyl haptomer is equivalent to an η^1 -C-bound iridium enolate.^[12] The veracity of this analysis is borne out by the following experiments. Upon exposure of the γ -acyloxy crotonate **1** to primary alcohol **2a** in the presence of the chiral *ortho*-cyclometalated iridium C,O-benzoate complex modified by (*R*)-Cl,MeO-biphep, (2,2'-bis(diphenylphosphino)-5,5'-dichloro-6,6'-dimethoxy-1,1'-biphenyl), designated (*R*)-**I**, the products of carbonyl addition were obtained in a combined 87 % yield as a 1:2 mixture of linear (88 % *ee*) and branched regioisomers, thus favoring formation of the branched adduct, which appeared as a roughly equimolar mixture of diastereomers (Table 1, entry 1). Catalytic C–C coupling of cyclopentanemethanol **2b** or cyclohexanemetha-

Table 2: Enantioselective vinylogous Reformatsky reaction from the aldehyde oxidation level.^[a]


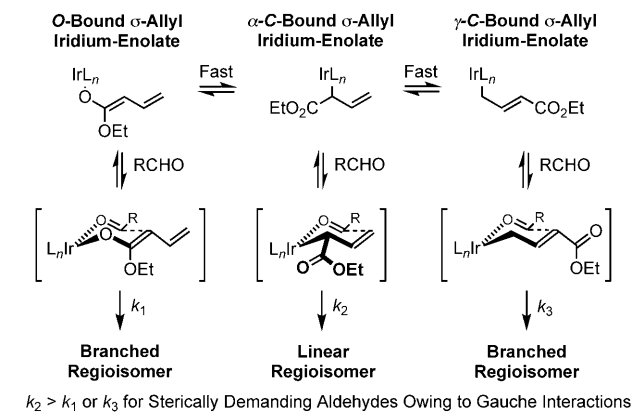
Entry	Product	Yield [%]	ee [%]	γ/α
1		65	83	1:2 ^[b]
2		80	98	6:1
3		88	92	6:1
4		85	96	$\geq 20:1$
5		70	99	$\geq 20:1$
6		83	99	$\geq 20:1$
7		60	93	$\geq 20:1$
8		58	93	$\geq 20:1$
9		80	94	$\geq 20:1$
10		94	96	5:1

[a] As described in Table 1.

nol **2c** under identical conditions revealed an increased proportion of the linear regioisomer (approximately 5:1 r.r. in each case), again accompanied by high levels of enantioselectivity (94 % *ee* and 95 % *ee*, respectively; Table 1, entries 2 and 3). Primary neopentyl alcohols **2d–2h** delivered products of C–C coupling with complete linear regioselectivity and exceptional enantioselectivity (Table 1, entries 4–8). In all cases, roughly equivalent yields and selectivities are observed when the C–C coupling was performed using aldehydes **3a–3h** (Table 2, entries 1–8). Finally, benzylic alcohols **2i** and **2j** and the corresponding aldehydes **3i** and **3j** engage in C–C coupling to furnish adducts **4i** and **4j** (Table 1, entries 9 and 10 and Table 2, entries 9 and 10, respectively). A detailed description of the assignment of absolute stereochemistry is provided in the Supporting Information.

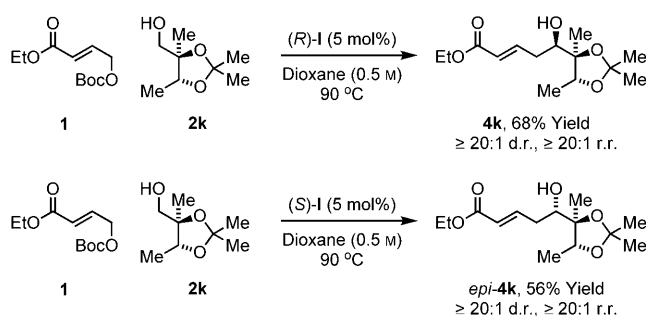
Linear regioselectivity in response to increased steric demand of the aldehyde suggests a Curtin–Hammett scenario wherein carbonyl addition occurs selectively from an equilibrating mixture of primary and secondary σ -allyl haptomer

mers.^[13,14] Beyond electronic effects, it is likely that carbonyl addition from the α -C-bound iridium enolate to form the less-substituted C–C bond is favored owing to the absence of gauche interactions in the transition state. For sterically demanding aldehydes, such gauche interactions would be accentuated, thus increasing the preference for the linear regioisomer (Scheme 3).



Scheme 3. Partitioning of branched and linear regioisomers.

To further explore the scope of the vinylogous Reformatsky–aldol addition, catalyst-directed diastereoselectivity was explored in the coupling of allylic carbonate **1** to α -chiral alcohol **2k** (Scheme 4).^[15] Under standard conditions employ-



Scheme 4. Catalyst-directed diastereoselectivity in vinylogous Reformatsky–aldol addition from the alcohol oxidation level.

ing (*R*)-**I** as the precatalyst, the linear adduct **4k** was isolated as a single regio- and stereoisomer. Using the enantiomeric precatalyst (*S*)-**I**, the linear adduct *epi*-**4k** was isolated as a single regio- and stereoisomer. Thus, complete levels of catalyst-directed diastereoselectivity are observed.^[16] In contrast, using the analogous achiral iridium complex modified by biphep, (2,2'-bis(diphenylphosphino)biphenyl), diastereoisomers **4k** and *epi*-**4k** were produced in 58% yield in a 1:1.5 ratio, respectively, along with a 9% yield of the branched regioisomer. It should be noted that although biphep is considered an achiral ligand, metal complexes of biphep adopt chiral racemic conformations that would attenuate substrate bias.

In summary, a catalytic enantioselective vinylogous Reformatsky-type addition has been described in which

asymmetric carbonyl addition occurs with equal facility from the alcohol or aldehyde oxidation level. Good to excellent levels of regioselectivity and uniformly high levels of enantioselectivity are observed across a range of alcohols **2a–2i** and aldehydes **3a–3i**. Furthermore, as demonstrated in the case of α -chiral alcohol **2k**, exceptional levels of catalyst-directed diastereoselectivity may be achieved. Insight into the structure–interaction features of the catalytic system vis-à-vis partitioning of linear and branched adducts suggests a Curtin–Hammett scenario, wherein carbonyl addition occurs selectively from an equilibrating mixture of primary and secondary α -allyl haptomers.^[13,14] The collective data are consistent with carbonyl addition from the secondary α -C-bound iridium enolate to form the less-substituted C–C bond owing to the absence of gauche interactions in the transition state. Notably, in reactions conducted from the alcohol oxidation level, the only stoichiometric by-products formed are carbon dioxide and *tert*-butanol. Future studies will focus on the development of related C–C bond-forming processes that combine oxidation/construction event and thus bypass discrete alcohol oxidation and stoichiometric organometallic reagents while gaining access to more tractable synthetic intermediates in the form of alcohols.

Received: January 25, 2011

Published online: March 4, 2011

Keywords: aldol reaction · allylation · homogeneous catalysis · iridium · transfer hydrogenation

[1] R. C. Fuson, *Chem. Rev.* **1935**, *16*, 1–27.

[2] For selected reviews on the vinylogous aldol reaction, see: a) G. Casiraghi, F. Zanardi, *Chem. Rev.* **2000**, *100*, 1929–1972; b) A. Soriente, M. De Rosa, R. Villano, A. Scettri, *Curr. Org. Chem.* **2004**, *8*, 993–1007; c) S. E. Denmark, J. R. Heemstra, Jr., G. L. Beutner, *Angew. Chem.* **2005**, *117*, 4760–4777; *Angew. Chem. Int. Ed.* **2005**, *44*, 4682–4698; d) M. Kalesse, *Top. Curr. Chem.* **2005**, *244*, 43–76; e) S. Hosokawa, K. Tatsuta, *Mini-Rev. Org. Chem.* **2008**, *5*, 1–18; f) F. Zanardi, L. Battistini, C. Curti, G. Casiraghi, G. Rassu, *Chemtracts: Org. Chem.* **2010**, *23*, 123–140.

[3] For examples of enantioselective vinylogous aldol reactions of β -ketoester derived dienols and dienolates, see: a) M. Sato, S. Sunami, Y. Sugita, C. Kaneko, *Chem. Pharm. Bull.* **1994**, *42*, 839–845; b) R. A. Singer, E. M. Carreira, *J. Am. Chem. Soc.* **1995**, *117*, 12360–12361; c) M. Sato, S. Sunami, Y. Sugita, C. Kaneko, *Heterocycles* **1995**, *41*, 1435–1444; d) M. De Rosa, M. R. Acocella, R. Villano, A. Soriente, A. Scettri, *Tetrahedron Lett.* **2003**, *44*, 6087–6090; e) R. Villano, M. R. Acocella, M. De Rosa, A. Soriente, A. Scettri, *Tetrahedron: Asymmetry* **2004**, *15*, 2421–2424; f) D. A. Evans, J. A. Murry, M. C. Kozlowski, *J. Am. Chem. Soc.* **1996**, *118*, 5814–5815; g) D. A. Evans, M. C. Kozlowski, J. A. Murry, C. S. Burgey, K. R. Campos, B. T. Connell, R. J. Staples, *J. Am. Chem. Soc.* **1999**, *121*, 669–685; h) J. Krüger, E. M. Carreira, *J. Am. Chem. Soc.* **1998**, *120*, 837–838; i) B. L. Pagenkopf, J. Krüger, A. Stojanovic, E. M. Carreira, *Angew. Chem.* **1998**, *110*, 3312–3314; *Angew. Chem. Int. Ed.* **1998**, *37*, 3124–3126; j) Y. Shimada, Y. Matsuoka, R. Irie, T. Katsuki, *Synlett* **2004**, 57–60; k) S. E. Denmark, G. L. Beutner, *J. Am. Chem. Soc.* **2003**, *125*, 7800–7801; l) S. E. Denmark, G. L. Beutner, T. Wynn, M. D. Eastgate, *J. Am. Chem. Soc.* **2005**, *127*, 3774–3789; m) S. E. Denmark, J. R. Heemstra, Jr., *J. Org. Chem.* **2007**, *72*, 5668–5686; n) V. B. Gondí, M. Gravel, V. H.

- Rawal, *Org. Lett.* **2005**, *7*, 5657–5660; o) T. Yoshinari, K. Ohmori, K. Suzuki, *Chem. Lett.* **2010**, *39*, 1042–1044.
- [4] For examples of enantioselective vinylogous aldol reactions of dienols and dienolates derived from simple α,β -unsaturated carbonyl compounds, see: a) G. Bluet, J. M. Campagne, *Tetrahedron Lett.* **1999**, *40*, 5507–5509; b) G. Bluet, J. M. Campagne, *J. Org. Chem.* **2001**, *66*, 4293–4298; c) B. Bazán-Tejeda, G. Bluet, G. Broustal, J. M. Campagne, *Chem. Eur. J.* **2006**, *12*, 8358–8366; d) S. E. Denmark, J. R. Heemstra, Jr., *Synlett* **2004**, 2411–2416; e) X. Moreau, B. Bazán-Tejeda, J. M. Campagne, *J. Am. Chem. Soc.* **2005**, *127*, 7288–7289; f) S. E. Denmark, J. R. Heemstra, Jr., *J. Am. Chem. Soc.* **2006**, *128*, 1038–1039; g) P. Rémy, M. Langner, C. Bolm, *Org. Lett.* **2006**, *8*, 1209–1211; h) L. V. Heumann, G. E. Keck, *Org. Lett.* **2007**, *9*, 4275–4278; i) S. Simsek, M. Horzella, M. Kalesse, *Org. Lett.* **2007**, *9*, 5637–5639; j) M. Fings, D. Goedert, C. Bolm, *Chem. Commun.* **2010**, *46*, 5497–5499; k) L. Ratjen, P. Garcia-Garcia, F. Lay, M. E. Beck, B. List, *Angew. Chem.* **2011**, *123*, 780–784; *Angew. Chem. Int. Ed.* **2011**, *50*, 754–758.
- [5] For examples of enantioselective vinylogous aldol reactions of 2-siloxy furan and 2-siloxy pyrrole, see: a) M. Szlosek, X. Franck, B. Figadère, A. Cavé, *J. Org. Chem.* **1998**, *63*, 5169–5172; b) Y. Matsuoka, R. Irie, T. Katsuki, *Chem. Lett.* **2003**, *32*, 584–585; c) S. Onitsuka, Y. Matsuoka, R. Irie, T. Katsuki, *Chem. Lett.* **2003**, *32*, 974–975; d) H. Nagao, Y. Yamane, T. Mukaiyama, *Chem. Lett.* **2007**, *36*, 8–9; e) M. Frings, I. Atodiresei, J. Runsink, G. Raabe, C. Bolm, *Chem. Eur. J.* **2009**, *15*, 1566–1569; f) C. Curti, A. Sartori, L. Battistini, G. Rassu, F. Zanardi, G. Casiraghi, *Tetrahedron Lett.* **2009**, *50*, 3428–3431; g) R. P. Singh, B. M. Foxman, L. Deng, *J. Am. Chem. Soc.* **2010**, *132*, 9558–9560.
- [6] For “direct” enantioselective vinylogous aldol additions of unsaturated carbonyl compounds and nitriles, see: a) H. Ube, N. Shimada, M. Terada, *Angew. Chem.* **2010**, *122*, 1902–1905; *Angew. Chem. Int. Ed.* **2010**, *49*, 1858–1861; b) Y. Yang, K. Zheng, J. Zhao, J. Shi, L. Lin, X. Liu, X. Feng, *J. Org. Chem.* **2010**, *75*, 5382–5384; c) R. Yazaki, N. Kumagai, M. Shibasaki, *J. Am. Chem. Soc.* **2010**, *132*, 5522–5531.
- [7] For a related vinylogous aldol-oxa-Michael cascade reaction, see: K. Liu, W. D. Woggon, *Eur. J. Org. Chem.* **2010**, 1033–1036.
- [8] D. Zhao, K. Oisaki, M. Kanai, M. Shibasaki, *J. Am. Chem. Soc.* **2006**, *128*, 14440–14441.
- [9] For the seminal report of the vinylogous Reformatsky-type reaction, see: a) R. C. Fuson, R. T. Arnold, H. G. Cooke, Jr., *J. Am. Chem. Soc.* **1938**, *60*, 2272–2273; b) R. C. Fuson, P. L. Southwick, *J. Am. Chem. Soc.* **1944**, *66*, 679–681.
- [10] For enantioselective carbonyl allylation by iridium-catalyzed C–C bond-forming transfer hydrogenation and related processes, see: a) I. S. Kim, M.-Y. Ngai, M. J. Krische, *J. Am. Chem. Soc.* **2008**, *130*, 6340–6341; b) I. S. Kim, M.-Y. Ngai, M. J. Krische, *J. Am. Chem. Soc.* **2008**, *130*, 14891–14899; c) I. S. Kim, S.-B. Han, M. J. Krische, *J. Am. Chem. Soc.* **2009**, *131*, 2514–2520; d) S. B. Han, I.-S. Kim, H. Han, M. J. Krische, *J. Am. Chem. Soc.* **2009**, *131*, 6916–6917; e) Y. Lu, I.-S. Kim, A. Hassan, D. J. Del Valle, M. J. Krische, *Angew. Chem.* **2009**, *121*, 5118–5121; *Angew. Chem. Int. Ed.* **2009**, *48*, 5018–5021; f) J. Itoh, S. B. Han, M. J. Krische, *Angew. Chem.* **2009**, *121*, 6431–6434; *Angew. Chem. Int. Ed.* **2009**, *48*, 6313–6316; g) Y. Lu, M. J. Krische, *Org. Lett.* **2009**, *11*, 3108–3111; h) A. Hassan, Y. Lu, M. J. Krische, *Org. Lett.* **2009**, *11*, 3112–3115; i) B. Bechem, R. L. Patman, S. Hashmi, M. J. Krische, *J. Org. Chem.* **2010**, *75*, 1795–1798; j) S. B. Han, H. Han, M. J. Krische, *J. Am. Chem. Soc.* **2010**, *132*, 1760–1761; k) Y. J. Zhang, J. H. Yang, S. H. Kim, M. J. Krische, *J. Am. Chem. Soc.* **2010**, *132*, 4562–4563; l) S. B. Han, X. Gao, M. J. Krische, *J. Am. Chem. Soc.* **2010**, *132*, 9153–9156; m) J. R. Zbieg, T. Fukuzumi, M. J. Krische, *Adv. Synth. Catal.* **2010**, *352*, 2416–2420; for recent applications in total synthesis, see: n) P. Harsh, G. A. O'Doherty, *Tetrahedron* **2009**, *65*, 5051–5055; o) S. B. Han, A. Hassan, I.-S. Kim, M. J. Krische, *J. Am. Chem. Soc.* **2010**, *132*, 15559–15561; p) P. Sawant, M. E. Maier, *Tetrahedron* **2010**, *66*, 9738–9744.
- [11] For selected reviews on C–C bond-forming hydrogenation and transfer hydrogenation, see: a) M.-Y. Ngai, J. R. Kong, M. J. Krische, *J. Org. Chem.* **2007**, *72*, 1063–1072; b) E. Skucas, M.-Y. Ngai, V. Komanduri, M. J. Krische, *Acc. Chem. Res.* **2007**, *40*, 1394–1410; c) F. Shibahara, M. J. Krische, *Chem. Lett.* **2008**, *37*, 1102–1107; d) R. L. Patman, J. F. Bower, I. S. Kim, M. J. Krische, *Aldrichim. Acta* **2008**, *41*, 95–104; e) J. F. Bower, I. S. Kim, R. L. Patman, M. J. Krische, *Angew. Chem.* **2009**, *121*, 36–48; *Angew. Chem. Int. Ed.* **2009**, *48*, 34–46; f) S. B. Han, I. S. Kim, M. J. Krische, *Chem. Commun.* **2009**, 7278–7287; g) J. F. Bower, M. J. Krische, *Top. Organomet. Chem.* **2011**, *43*, 107–138.
- [12] For reviews covering the coordination chemistry of late-transition-metal enolates, including their η^1 -C-bound haptomers, see: a) E. R. Burkhardt, J. J. Doney, G. A. Sough, J. M. Stack, C. H. Heathcock, R. G. Bergman, *Pure Appl. Chem.* **1988**, *60*, 1–6; b) J. Vicente in *The Chemistry Of Metal Enolates* (Ed.: J. Zabicky), Wiley, Chichester, **2009**, pp. 313–353.
- [13] Such Curtin–Hammett effects appear evident in Nozaki–Hiyama crotylation reactions: a) T. Hiyama, Y. Okude, K. Kimura, H. Nozaki, *Bull. Chem. Soc. Jpn.* **1982**, *55*, 561–568; b) W. R. Roush in *Comprehensive Organic Synthesis*, Vol. 2 (Eds.: B. M. Trost, I. Fleming, C. H. Heathcock), Pergamon, Oxford, **1991**, pp. 1–53.
- [14] For Curtin–Hammett effects in ruthenium-catalyzed carbonyl addition reactions involving 1,1-disubstituted allenes as allyl donors, see: J. R. Zbieg, E. L. McInturff, J. C. Leung, *J. Am. Chem. Soc.* **2011**, *133*, 1141–1144.
- [15] Alcohol **2k** was prepared by Sharpless asymmetric dihydroxylation according to the following literature procedure: B. J. D. Wright, J. Hartung, F. Peng, R. Van der Water, H. Liu, Q.-H. Tan, T.-C. Chou, S. J. Danishefsky, *J. Am. Chem. Soc.* **2008**, *130*, 16786–16790.
- [16] For selected examples of catalyst-directed diastereoselectivity, see: a) N. Minami, S. S. Ko, Y. Kishi, *J. Am. Chem. Soc.* **1982**, *104*, 1109–1111; b) S. Y. Ko, A. W. M. Lee, S. Masamune, L. A. Reed III, K. B. Sharpless, F. J. Walker, *Science* **1983**, *220*, 949–951; c) S. Kobayashi, A. Ohtsubo, T. Mukaiyama, *Chem. Lett.* **1991**, 831–834; d) A. Hammadi, J. M. Nuzillard, J. C. Poulin, H. B. Kagan, *Tetrahedron: Asymmetry* **1992**, *3*, 1247–1262; e) M. P. Doyle, A. V. Kalinin, D. G. Ene, *J. Am. Chem. Soc.* **1996**, *118*, 8837–8846; f) B. M. Trost, T. L. Calkins, C. Oertelt, J. Zambrano, *Tetrahedron Lett.* **1998**, *39*, 1713–1716; g) E. P. Balskus, E. N. Jacobsen, *Science* **2007**, *317*, 1736–1740; h) S. B. Han, J. R. Kong, M. J. Krische, *Org. Lett.* **2008**, *10*, 4133–4135.